# An Expeditious Synthesis of 1,2,3,4-Tetrahydro-1,1-dimethylphenanthrenes 

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#### Abstract

A simple convergent and general method has been developed for the synthesis of 1,2,3,4-tetrahydro-1,1-dimethylphenanthrenes 4a-f and a few 1,2,3,4,9,10-hexahydro-1,1-dimethylphenanthrenes 5c-f by polyphosphoric acid catalysed reaction of the easily accessible 2-(2-arylethyl)-3,3-dimethylcyclohexanones $3 \mathbf{a}-\mathbf{f}$. The tetrahydrophenanthrenes $4 \mathbf{a}, \mathbf{c}, \mathbf{e}, \mathbf{f}$ have been converted to the respective benzylic ketones $\mathbf{6 a , c}, \mathbf{e}$,f by oxidation with pyridinium chlorochromate and Celite.


The 1,1-dimethyltetrahydrophenanthrenes $\mathbf{4 a}{ }^{1,2}$ and $\mathbf{4 b}{ }^{\mathbf{3}}$ and $\mathbf{4 c},{ }^{4}$ key intermediates used in a number of the total syntheses of miltirone $1^{5}$ and tanshinone IIA ${ }^{6} 2$, respectively, two important members of a large group of highly bio-active abietane diterpenoid quinones ${ }^{2}$ isolated from the roots of Salvia miltiorrhiza Bunge (Danshen), have been prepared by a lengthy sequence of reactions from substituted benzene and naphthalene derivatives. We describe now a simple convergent and a highly efficient general synthetic route to $\mathbf{4 a - c}$, simonellite $\mathbf{4 d}^{7}$ and the related tetrahydrophenanthrenes $\mathbf{4 e}, \mathbf{f}$ through the easily accessible cyclohexanone derivatives $\mathbf{3 a}-\mathbf{f}$, used in our recent synthesis ${ }^{8} 12$ of some diterpenoids.


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4a, 6a;
$R^{1}=H, R^{2}=\mathrm{Pr}^{i}, R^{3}=\mathrm{OMe}$
$3 b, 4 b ; \quad R^{1}=R^{3}=O M e, R^{2}=H$
$3 c, 4 c, 5 c, 6 c ; R^{1}=R^{2}=H, R^{3}=O M e$
3d, 4d, 5d; $\quad R^{1}=R^{3}=H, R^{2}=P r^{i}$
3e, 4e, 5e, 6e; $R^{1}=R^{3}=H, R^{2}=O M e$
$3 f, 4 f, 5 f, 6 f ; \quad R^{1}=R^{2}=R^{3}=H$
$\dagger$ Since the reaction was carried out under $\mathrm{N}_{2}$ the dehydrogenation during the cyclisations possibly involved oxygen dissolved in the reaction medium. However, for a given substrate, the yield of the tetrahydro product does not parallel the length of the reaction in PPA.

Polyphosphoric acid induced reactions of $\mathbf{3 a}{ }^{8}$ and $\mathbf{3 b}{ }^{\mathbf{1 2}}$ in boiling toluene directly afforded $\mathbf{4} \mathbf{a}^{1,2}$ and $\mathbf{4 b},{ }^{3}$ in 75 and $79 \%$ yields respectively, as the only isolable products by concomitant cyclodehydration and aromatisation. ${ }^{13}$ Under the similar conditions, however, the ketones $\mathbf{3 c},{ }^{9} \mathbf{3 d},{ }^{11} \mathbf{3 e}{ }^{10}$ and $\mathbf{3 f}{ }^{9}$ gave the respective tetrahydrophenanthrenes $4 \mathbf{c},{ }^{3} 4 \mathbf{d},{ }^{7} 4 \mathbf{e}^{14}$ and $4 f$ in high yields $\dagger$ along with minor amounts of the corresponding hexahydrophenanthrenes $5 \mathbf{c}-\mathbf{f}$, separated by column chromatography. Each of these mixtures on dehydrogenation directly with palladium-charcoal in boiling xylene led to the corresponding tetrahydrophenanthrenes in $82-95 \%$ yields.

The ketones $6 a^{2,15}$ and $6 \mathbf{e},{ }^{16}$ prepared by a lengthy sequence of reactions, have been transformed recently to some diterpenoids by a reductive angular methylation reaction. ${ }^{15,16}$ We have found that the easily accessible tetrahydrophenanthrenes $\mathbf{4 a}, \mathbf{4 c}, \mathbf{4 e}$ and $\mathbf{4 f}$ undergo smooth benzylic oxidation ${ }^{17}$ with pyridinium chlorochromate (PCC)-Celite in dichloromethane to afford the respective ketones $6 \mathbf{6}, \mathbf{6 c}$, $\mathbf{6 e}$ and $\mathbf{6 f}$ in $82-85 \%$ yields.

In conclusion, in the present work a simple convergent and general synthetic route has been developed for some key hydrophenanthrene intermediates for the synthesis of diterpenoids.

## Experimental

IR spectra of solids ( K Br ) and liquids (film) were recorded on a Perkin-Elmer model PE 298 instrument. UV spectra were recorded on a Beckman DU spectrometer for solutions in ethanol $(95 \%)$. ${ }^{1} \mathrm{H}$ NMR spectra were recorded at 200 MHz on an XL-200 spectrometer for solutions in $\mathrm{CDCl}_{3}$ with $\mathrm{SiMe}_{4}$ as internal standard, $J$ values are given in Hz . Analytical GLC was performed on a Shimadzu GC-9A model with a flame-ionisation detector employing a $1.5 \% \mathrm{OV}-17(6.5 \mathrm{ft} . \times 0.25 \mathrm{in})$ column with $\mathrm{N}_{2}$ as the carrier gas. Column chromatography was performed on neutral alumina (Brockmann Grade 1, of BDH, India) or silica gel [Glaxo Laboratories (India) Ltd.]. Light petroleum refers to the fraction of b.p. $40-60^{\circ} \mathrm{C}$ unless otherwise stated. Ether refers to diethyl ether. Elemental analyses were performed by Mr. P. P. Bhattacharya and S. K. Sarkar of this laboratory.

Cyclisation of 3a to 4a.-To a well-stirred mixture of polyphosphoric acid (PPA) prepared from orthophosphoric acid ( $15 \mathrm{~cm}^{3}, 85 \%$ ) and phosphorus pentoxide ( 23 g ) at $115-$ $120^{\circ} \mathrm{C}$ (bath temp.) a solution of ketone $3 \mathbf{a}^{8}(1 \mathrm{~g}, 3.30 \mathrm{mmol})$ in toluene ( $10 \mathrm{~cm}^{3}$ ) was added, stirring at the same temperature was continued for 10 h . The red mixture was decomposed with crushed ice and the organic matter was extracted with ether. The ethereal extracts were washed thoroughly with water and $3 \%$ aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}$ and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. Removal of the solvent afforded 1,2,3,4-tetrahydro-7-isopropyl-6-methoxy-1,1-
dimethylphenanthrene $\mathbf{4 a}$ ( $700 \mathrm{mg}, 75 \%$ ), as a white solid, m.p. $83-84^{\circ} \mathrm{C}$ (from methanol) (lit.,' m.p. 83-85 ${ }^{\circ} \mathrm{C}$ ); $v_{\text {max }} / \mathrm{cm}^{-1} 1627$ and 1605; $\lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 238(\log \varepsilon 4.92) ; \delta 1.30(6 \mathrm{H}, \mathrm{d}, J$ 7, CHMe $e_{2}$, $1.35\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 1.70-1.79\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right), 1.92-$ $2.04\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}\right), 3.07\left(2 \mathrm{H}, \mathrm{t}, J 7,4-\mathrm{H}_{2}\right), 3.35-3.50(1 \mathrm{H}, \mathrm{m}$, CHMe ${ }_{2}$ ), 3.98 ( $3 \mathrm{H}, \mathrm{s}$, ArOMe), 7.22 ( $1 \mathrm{H}, \mathrm{s}, 5-\mathrm{ArH}$ ), $7.40(1 \mathrm{H}, \mathrm{d}$, $J 8,10-\mathrm{ArH}$ ) and $7.66-7.75(2 \mathrm{H}, \mathrm{m}, 8-\mathrm{ArH}$ and $9-\mathrm{ArH})\left[\right.$ lit., ${ }^{1}$ $\delta\left(\mathrm{CDCl}_{3} ; 60 \mathrm{MHz}\right) 2.5-3.0(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 6.15(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, $6.70\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH} \mathrm{Me}_{2}\right), 7.05\left(2 \mathrm{H}, \mathrm{t}, J 6\right.$, benzylic $\left.\mathrm{CH}_{2}\right), 8.25$ $\left(4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{CH}_{2}\right)$ and 8.7-8.8 (12 H, m, $\left.4 \times \mathrm{Me}\right)$ ].

Cyclisation of $\mathbf{3 b}$ to $\mathbf{4 b}$. The ketone $\mathbf{3 b}{ }^{12}(1 \mathrm{~g}, 3.44 \mathrm{mmol})$ was converted in the same way as described for 4 a into $1,2,3,4$ -tetrahydro-5,7-dimethoxy-1,1-dimethylphenanthrene $\mathbf{4 b}$ which was obtained as a white solid ( $735 \mathrm{mg}, 79 \%$ ), m.p. $159-160^{\circ} \mathrm{C}$ (light petroleum) (lit., ${ }^{3}$ m.p. $158{ }^{\circ} \mathrm{C}$ ); $v_{\text {max }} / \mathrm{cm}^{-1} 1626$ and 1600 ; $\lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 238(\log \varepsilon 4.84)\left[\mathrm{lit}.{ }^{3}{ }^{3} \lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 238\right.$, 284, 295, 331)]; $\delta 1.34\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 1.64-1.78(2 \mathrm{H}, \mathrm{m}$, $\left.2-\mathrm{H}_{2}\right), 1.90-2.02\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}\right), 3.02\left(2 \mathrm{H}, \mathrm{t}, J 7,4-\mathrm{H}_{2}\right), 3.94(3 \mathrm{H}$, s, ArOMe), 3.96 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{ArOMe}$ ), 6.50 ( $1 \mathrm{H}, \mathrm{brs}, 5-\mathrm{ArH}$ ), 6.84 ( 1 $\mathrm{H}, \mathrm{br} \mathrm{s}, 7-\mathrm{ArH}), 7.38(1 \mathrm{H}, \mathrm{d}, J 8,10-\mathrm{ArH})$ and $8.0(1 \mathrm{H}, \mathrm{d}, J 8$, 9-ArH).

Cyclisation of 3c to 4c and 1,2,3,4,9,10-Hexahydro-6-methoxy-1,1-dimethylphenanthrene 5 c .-To a well stirred mixture of polyphosphoric acid (PPA) prepared from orthophosphoric acid ( $15 \mathrm{~cm}^{3}, 85 \%$ ) and phosphorus pentoxide ( 23 g ) at $115-$ $120{ }^{\circ} \mathrm{C}$ (bath temp.) a solution of ketone $3 c^{9}(1 \mathrm{~g}, 3.84 \mathrm{mmol})$ in toluene ( $10 \mathrm{~cm}^{3}$ ) was added, stirring at the same temperature was continued for 10 h . The red mixture was decomposed with crushed ice and the organic matter was extracted with ether. The ethereal extracts were washed thoroughly with water, $3 \%$ aqueous $\mathrm{Na}_{2} \mathrm{CO}_{3}$ and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. Removal of the solvent afforded a light yellow viscous liquid mixture of $4 \mathbf{c}$ and $5 \mathbf{c}$, which was chromatographed over neutral alumina ( 32 g ) and eluted with light petroleum ( $7 \times 35 \mathrm{~cm}^{3}$ ) to afford the title compound $5 \mathrm{c}(79 \mathrm{mg}, 8.5 \%)$, b.p. $150-155^{\circ} \mathrm{C}(1 \mathrm{mmHg})$ (Found: C, 83.95 ; $\mathrm{H}, 8.85 . \mathrm{C}_{17} \mathrm{H}_{22} \mathrm{O}$ requires $\mathrm{C}, 84.24 ; \mathrm{H}, 9.15 \%$ ); $v_{\text {max }} / \mathrm{cm}^{-1}$ $1605 ; \lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 206(\log \varepsilon 4.30)$ and $270(\log \varepsilon$ 4.20); $\delta 1.0\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right)$, $1.46-1.79(10 \mathrm{H}, \mathrm{m}), 3.73(3 \mathrm{H}, \mathrm{s}$, ArOMe) and 6.39-6.87 ( $3 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ).

Further elution with light petroleum $\left(60-80^{\circ} \mathrm{C}\right)\left(7 \times 30 \mathrm{~cm}^{3}\right)$ furnished 1,2,3,4-tetrahydro-6-methoxy-1,1-dimethylphenanthrene $\mathbf{4 c}(803 \mathrm{mg}, 87 \%)$ as a colourless oil which solidified on
 $\left.117-118.5{ }^{\circ} \mathrm{C}\right) ; v_{\text {max }} / \mathrm{cm}^{-1} 1625$ and $1600 ; \lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 236$ $(\log \varepsilon 4.54), 280(\log \varepsilon 3.64), 315(\log \varepsilon 3.08)$ and $330(\log \varepsilon$ $3.19)\left[\text { lit. }{ }^{4}\right\rangle_{\text {max }}($ EtOH $) / \mathrm{nm} 237,270$ sh, 279,289 sh, 316 and 331 $(\log \varepsilon 4.87,3.68,3.73,3.64,3.18$ and 3.31$)] ; \delta 1.37(6 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CMe}_{2}\right), 1.71-1.80\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right), 1.92-2.03\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}\right), 3.08$ $\left(2 \mathrm{H}, \mathrm{t}, J 8,4-\mathrm{H}_{2}\right), 3.96(3 \mathrm{H}, \mathrm{s}$, ArOMe), $7.14(1 \mathrm{H}, \mathrm{dd}, J 8$ and 1 , $7-\mathrm{ArH}$ ), 7.28 ( $1 \mathrm{H}, \mathrm{br}$ s, $5-\mathrm{ArH}$ ), 7.41 ( $1 \mathrm{H}, \mathrm{d}, J 8,10-\mathrm{ArH}$ ), 7.64 $(1 \mathrm{H}, \mathrm{d}, J 8,8-\mathrm{ArH})$ and $7.73(1 \mathrm{H}, \mathrm{d}, J 8,9-\mathrm{ArH})$.

Conversion of $\mathbf{3 c}$ to $\mathbf{4 c}$. The crude mixture of $\mathbf{4 c}$ and $5 \mathbf{c}$ prepared by cyclisation of the cyclohexanone $3 \mathrm{c}(250 \mathrm{mg}, 0.96$ mmol ) with PPA following the same procedure as described above, was dissolved in xylene ( $12 \mathrm{~cm}^{3}$ ) and mixed with $\mathrm{Pd}-\mathrm{C}$ $(10 \%)(125 \mathrm{mg})$. The magnetically stirred mixture was refluxed for 7 h . The reaction mixture was cooled, filtered and the residue was rinsed with ether $\left(25 \mathrm{~cm}^{3}\right)$. The ether washings and the filtrate were combined and the solvent was evaporated under reduced pressure. The resulting product was purified by chromatography on neutral alumina ( 10 g ) using light petroleum as eluent to afford $\mathbf{4 c}(217 \mathrm{mg}, 95 \%$ ); identical (m.p., mixed m.p., IR and ${ }^{1} \mathrm{H}$ NMR spectra and GLC) with the sample described above.
Cyclisation of 3d to 4d and 1,2,3,4,9,10-Hexahydro-7-iso-propyl-1,1-dimethylphenanthrene 5d.-The ketone $\mathbf{3 d}^{11}(1 \mathrm{~g}$, 3.67 mmol ) was converted in the same way as described for 4 a
into the mixture of $\mathbf{4 d}$ and $\mathbf{5 d}$. The mixture was chromatographed over neutral alumina ( 32 g ) and elution with light petroleum ( $7 \times 30 \mathrm{~cm}^{3}$ ) afforded the title compound $5 \mathrm{~d}(299 \mathrm{mg}, 32 \%$ ), b.p. $150-155^{\circ} \mathrm{C}(1 \mathrm{mmHg})$ (Found: C, $89.85 ; \mathrm{H}, 9.95 . \mathrm{C}_{19} \mathrm{H}_{26}$ requires C, $89.70 ; \mathrm{H}, 10.30 \%$ ); $v_{\text {max }} / \mathrm{cm}^{-1} 1605 ; i_{\text {max }}(\mathrm{EtOH}) /$ $\mathrm{nm} 214(\log \varepsilon 4.37), 221(\log \varepsilon 4.35)$ and $271(\log \varepsilon 4.21) ; \delta$ $1.08\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 1.24\left(6 \mathrm{H}, \mathrm{d}, J 6, \mathrm{CH} M e_{2}\right), 1.50-1.59(2 \mathrm{H}, \mathrm{m}$, $2-\mathrm{H}_{2}$ ), $1.70-1.84\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}\right), 2.15-2.28\left(2 \mathrm{H}, \mathrm{m}, 10-\mathrm{H}_{2}\right), 2.32-$ $2.44\left(2 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{2}\right), 2.69\left(2 \mathrm{H}, \mathrm{t}, J 7,9-\mathrm{H}_{2}\right), 2.80-2.94(1 \mathrm{H}, \mathrm{m}$, CH $\mathrm{Me}_{2}$ ), $7.02(1 \mathrm{H}, \mathrm{brs}, 8-\mathrm{ArH}), 7.09(1 \mathrm{H}, \mathrm{br} \mathrm{d}, 6-\mathrm{ArH})$ and 7.19 ( $1 \mathrm{H}, \mathrm{d}, J 8,5-\mathrm{ArH}$ ).
Further elution with light petroleum $\left(60-80^{\circ} \mathrm{C}\right)\left(7 \times 35 \mathrm{~cm}^{3}\right)$ furnished 1,2,3,4-tetrahydro-7-isopropyl-1,1-dimethylphenanthrene $\mathbf{4 d}(537 \mathrm{mg}, 58 \%)$ as a colourless oil which solidified on standing, m.p. $59{ }^{\circ} \mathrm{C}$ (from ethanol) (lit., ${ }^{7}$ m.p. $58-59{ }^{\circ} \mathrm{C}$ ); $v_{\text {max }} / \mathrm{cm}^{-1} 1627$ and $1605 ; \lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 231(\log \varepsilon 4.91)$ and $279\left(\log\right.$ \& 3.79) [lit., ${ }^{7} i_{\max } 233,271$ sh, 278, 291 sh, 310, 317 and 324 ( $\varepsilon 24300,6400,3800,750,250$ and 880)]; $\delta 1.34$ ( 6 $\left.\mathrm{H}, \mathrm{d}, J 6, \mathrm{CH} M e_{2}\right), 1.37\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 1.70-1.80\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right)$, $1.90-2.02\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}\right), 3.03-3.16\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH} \mathrm{Me}_{2}\right.$ and $\left.4-\mathrm{H}_{2}\right)$, 7.40-7.52 ( $2 \mathrm{H}, \mathrm{m}, 6$ - and 8-ArH), $7.60-7.67(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{and}$ $9-\mathrm{ArH})$ and $7.94(1 \mathrm{H}, \mathrm{d}, J 9,10-\mathrm{ArH})$.

Conversion of $\mathbf{3 d}$ to $\mathbf{4 d}$. The crude mixture of $\mathbf{4 d}$ and $\mathbf{5 d}$ obtained from the cyclisation of $\mathbf{3 d}$ ( $250 \mathrm{mg}, 0.91 \mathrm{mmol}$ ) as described above was converted in the same way as described for 3c into 4 d ( $190 \mathrm{mg}, 82 \%$ ), identical (m.p., mixed m.p., IR and ${ }^{1} \mathrm{H}$ NMR spectra and GLC) with the sample described above.

Cyclisation of $\mathbf{3 e}$ to $\mathbf{4 e}$ and 1,2,3,4,9,10-Hexahydro-7-methoxy1,1 -dimethylphenanthrene 5 e . -The ketone $3 \mathrm{e}^{10}(1 \mathrm{~g}, 3.84 \mathrm{mmol})$ was converted in the same way as described for 3a into the mixture of 4 e and 5 e . The mixture was chromatographed over neutral alumina ( 32 g ) and elution with light petroleum ( $6 \times 35 \mathrm{ml}$ ) afforded $\mathbf{5 e}(232 \mathrm{mg}, 25 \%)$ as a colourless oil which solidified on standing, m.p. $49-50^{\circ} \mathrm{C}$ (from methanol) (Found: C, $83.95 ; \mathrm{H}, 9.35 . \mathrm{C}_{17} \mathrm{H}_{22} \mathrm{O}$ requires $\mathrm{C}, 84.24 ; \mathrm{H}, 9.15 \%$ ); $v_{\text {max }} / \mathrm{cm}^{-1} 1605 ; i_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 206(\log \varepsilon 4.30)$ and $272(\mathrm{log}$ $\varepsilon 4.20) ; \delta 1.07\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 1.37-2.67(10 \mathrm{H}, \mathrm{m}), 3.70(3 \mathrm{H}, \mathrm{s}$, ArOMe) and 6.40-7.0 ( $3 \mathrm{H}, \mathrm{m}, \mathrm{ArH}$ ).
Further elution with light petroleum $\left(60-80^{\circ} \mathrm{C}\right)\left(7 \times 30 \mathrm{~cm}^{3}\right)$ furnished $4 \mathrm{e}\left(572 \mathrm{mg}, 62 \%\right.$ ) as a colourless solid, m.p. $56-57^{\circ} \mathrm{C}$ (from methanol) (lit., ${ }^{14} \mathrm{~m} . \mathrm{p} .55 .5-56^{\circ} \mathrm{C}$ ); $v_{\max } / \mathrm{cm}^{-1} 1628$ and $1600 ; \lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 228(\log \varepsilon 4.87) ; \delta 1.35\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right)$, $1.68-1.78\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right), 1.88-2.02\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}\right), 3.11(2 \mathrm{H}, \mathrm{t}, \mathrm{J}$ $\left.8,4-\mathrm{H}_{2}\right), 3.93(3 \mathrm{H}, \mathrm{s}, \mathrm{ArOMe}), 7.12-7.26(2 \mathrm{H}, \mathrm{m}, 6-\mathrm{and} 8-\mathrm{ArH})$, $7.50(1 \mathrm{H}, \mathrm{d}, J 8,9-\mathrm{ArH}), 7.62(1 \mathrm{H}, \mathrm{d}, J 8,5-\mathrm{ArH})$ and $7.94(1 \mathrm{H}$, d, $J 8,10-\mathrm{ArH}$ ).

Conversion of $\mathbf{3 e}$ into 4 e . The crude mixture of $\mathbf{4 e}$ and 5 e obtained from the cyclisation of $3 \mathrm{e}(250 \mathrm{mg}, 0.96 \mathrm{mmol})$ was converted in the same way as described for $\mathbf{4 c}$ into $\mathbf{4 e}(196 \mathrm{mg}$, $85 \%$ ), identical (m.p., mixed m.p., IR and ${ }^{1} \mathrm{H}$ NMR spectra and GLC) with the sample described above.

1,2,3,4-Tetrahydro-1,1-dimethylphenanthrene $\mathbf{4 f}$ and 1,2,3,4,9,10-Hexahydro-1,1-dimethylphenanthrene $\quad \mathbf{5 f}$.-The ketone 3 f ${ }^{10}(1 \mathrm{~g}, 4.34 \mathrm{mmol})$ was converted in the same way as described for $\mathbf{4 a}$ into the mixture of $\mathbf{4 f}$ and $\mathbf{5 f}$. The mixture was chromatographed over neutral alumina ( 36 g ) and elution with light petroleum ( $10 \times 30 \mathrm{~cm}^{3}$ ) afforded the pure $\mathbf{5 f}(64 \mathrm{mg}, 7 \%$ ), b.p. $140-145^{\circ} \mathrm{C}(1 \mathrm{mmHg})$ (Found: C, 90.3; H, 9.65. $\mathrm{C}_{16} \mathrm{H}_{20}$ requires C, $90.50 ; \mathrm{H}, 9.50 \%)$; $v_{\text {max }} / \mathrm{cm}^{-1} 1605 ; \lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm}$ $207(\log \varepsilon 4.04), 213(\log \varepsilon 4.04), 220(\log \varepsilon 4.04)$ and $268(\log$ $\varepsilon 3.80) ; \delta 1.08\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 1.50-1.60\left(2 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}_{2}\right), 1.70-1.84$ ( $2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}$ ), 2.12-2.26 ( $2 \mathrm{H}, \mathrm{m}, 10-\mathrm{H}_{2}$ ), $2.44\left(2 \mathrm{H}, \mathrm{m}, 4-\mathrm{H}_{2}\right.$ ), $2.67\left(2 \mathrm{H}, \mathrm{t}, J 8,9-\mathrm{H}_{2}\right)$ and $7.10-7.28(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH})$.
Further elution with light petroleum $\left(60-80^{\circ} \mathrm{C}\right)\left(7 \times 35 \mathrm{~cm}^{3}\right)$ afforded $4 \mathrm{f}(819 \mathrm{mg}, 88 \%)$ as a colourless oil, b.p. $165-170^{\circ} \mathrm{C}$ $(1 \mathrm{mmHg})$ (Found: C, $91.1 ; \mathrm{H}, 8.4 . \mathrm{C}_{16} \mathrm{H}_{18}$ requires C, $91.37 ; \mathrm{H}$,
$8.63 \%) ; v_{\max } / \mathrm{cm}^{-1} 1627$ and $1600 ; \lambda_{\max }(\mathrm{EtOH}) / \mathrm{nm} 229(\log \varepsilon$ 4.48) and $281(\log \varepsilon 3.72) ; \delta 1.37\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 1.72-1.80(2 \mathrm{H}$, $\left.\mathrm{m}, 2-\mathrm{H}_{2}\right), 1.92-2.01\left(2 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}_{2}\right), 3.14\left(2 \mathrm{H}, \mathrm{t}, J 7,4-\mathrm{H}_{2}\right)$, $7.47-7.70(4 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 7.81(1 \mathrm{H}, \mathrm{d}, J 8,10-\mathrm{ArH})$ and $8.03(1 \mathrm{H}$, d, J8, 9-ArH).

Conversion of $\mathbf{3 f}$ to $\mathbf{4 f}$. The crude mixture of $\mathbf{4 f}$ and $\mathbf{5 f}$ obtained from the cyclisation of $\mathbf{3 f}(250 \mathrm{mg}, 1.08 \mathrm{mmol})$ as described above, was converted in the same way as described for $\mathbf{4 c}$ into $\mathbf{4 f}$ ( $217 \mathrm{mg}, 95 \%$ ), identical (IR, GLC and ${ }^{1} \mathrm{H}$ NMR spectra) with the sample described above.

Oxidation of $\mathbf{4 a}$ to $\mathbf{6 a}$.-To a solution of the phenanthrene $\mathbf{4 a}$ $(300 \mathrm{mg}, 1.06 \mathrm{mmol})$ in dichloromethane $(10 \mathrm{ml})$, was added a finely powdered and homogenised mixture of PCC $(1.1 \mathrm{~g}, 5.10$ $\mathrm{mmol})$ and Celite $(1.1 \mathrm{~g}) .{ }^{17 a}$ The reaction mixture was stirred for 25 h at room temperature and then diluted with ether ( $10 \mathrm{~cm}^{3}$ ) and filtered through a short pad of Celite and anhydrous magnesium sulphate. The filter cake was washed with two portions of ether $\left(10 \mathrm{~cm}^{3} \times 2\right)$ and the combined filtrate was evaporated under reduced pressure to afford $\mathbf{6 a}(258 \mathrm{mg}, 82 \%)$, as a colourless solid, m.p. $88-89^{\circ} \mathrm{C}$ (from methanol) (lit., ${ }^{2} \mathrm{~m} . \mathrm{p}$. 89-91 C): $v_{\text {max }} / \mathrm{cm}^{-1} 1672,1625$ and $1600 ; \lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 225$ $\left.(\log \varepsilon 4.50) ; \delta 1.30(6 \mathrm{H}, \mathrm{d}, J 7, \mathrm{CHMe})_{2}\right), 1.46\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right)$, $2.10\left(2 \mathrm{H}, \mathrm{t}, J 8,2-\mathrm{H}_{2}\right), 2.88\left(2 \mathrm{H}, \mathrm{t}, J 8,3-\mathrm{H}_{2}\right), 3.39-3.49(1 \mathrm{H}, \mathrm{m}$, CH Me ${ }_{2}$ ), $4.04(3 \mathrm{H}, \mathrm{s}, \mathrm{ArOMe}), 7.45(1 \mathrm{H}, \mathrm{d}, J 8,10-\mathrm{ArH}), 7.62$ ( $1 \mathrm{H}, \mathrm{s}, 8-\mathrm{ArH}$ ), $7.96(1 \mathrm{H}, \mathrm{d}, J 8,9-\mathrm{ArH})$ and $8.94(1 \mathrm{H}, \mathrm{s}, 5-\mathrm{ArH})$ [lit., ${ }^{2}{ }^{1} \mathrm{H}$ NMR: $\delta 1.29(\mathrm{~d}, J 7,6 \mathrm{H}), 1.45(\mathrm{~s}, 6 \mathrm{H}), 2.07(\mathrm{t}, J 7,2$ H), $2.84(\mathrm{t}, J 7,2 \mathrm{H}), 3.40$ (septet, $J 7,1 \mathrm{H}), 4.00(\mathrm{~s}, 3 \mathrm{H}), 7.38,7.80$ $\left(\mathrm{AB}_{\mathrm{q}}, J 9.2 \mathrm{H}\right), 7.55(\mathrm{~s}, 1 \mathrm{H})$ and $\left.8.78(\mathrm{~s}, 1 \mathrm{H})\right]$.

1,2,3,4-Tetrahydro-6-methoxy-1,1-dimethylphenanthrene-4one $6 \mathbf{6}$.-The compound $4 \mathrm{c}(300 \mathrm{mg}, 1.25 \mathrm{mmol})$ was converted, in the same way as described for $\mathbf{6 a}$ into the title compound $\mathbf{6 c}$ ( $266 \mathrm{mg}, 84^{\circ}{ }_{\circ}$ ), b.p. $165-170^{\circ} \mathrm{C}(0.1 \mathrm{mmHg})$ (Found: $\mathrm{C}, 80.5 ; \mathrm{H}$, 7.1. $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{O}_{2}$ requires $\mathrm{C}, 80.28 ; \mathrm{H}, 7.13 \%$ ); $v_{\max } / \mathrm{cm}^{-1} 1670$, 1626 and $1600: i_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 218(\log \varepsilon 4.50)$ and $247(\log \varepsilon$ $4.31) ; \delta 1.48\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 2.11\left(2 \mathrm{H}, \mathrm{t}, J 8,2-\mathrm{H}_{2}\right), 2.89(2 \mathrm{H}, \mathrm{t}$, $\left.J 8,3-\mathrm{H}_{2}\right), 4.01(3 \mathrm{H}, \mathrm{s}$, ArOMe), $7.21(1 \mathrm{H}, \mathrm{dd}, J 8$ and $1,7-\mathrm{ArH})$, $7.46(1 \mathrm{H}, \mathrm{d}, J 8,10-\mathrm{ArH}), 7.75(1 \mathrm{H}, \mathrm{d}, J 8,8-\mathrm{ArH}), 7.97(1 \mathrm{H}, \mathrm{d}, J$ $8,9-\mathrm{ArH})$ and $8.96(1 \mathrm{H}, \mathrm{d}, J 1,5-\mathrm{ArH})$.

Oxidation of $\mathbf{4 e}$ to $\mathbf{6 e}$.-The compound $\mathbf{4 e}(300 \mathrm{mg}, 1.25$ mmol ) was converted in the same way as described for $\mathbf{6 a}$ into $\mathbf{6 e}$ ( $269 \mathrm{mg}, 85 \%$ ), b.p. $160{ }^{\circ} \mathrm{C}\left(0.1 \mathrm{mmHg}\right.$ ) [lit., ${ }^{16}$ b.p. (bath temp.) $170 \mathrm{C} / 0.1 \mathrm{mmHg}] ; v_{\max } / \mathrm{cm}^{-1} 1673,1625$ and 1600 [lit., ${ }^{16}$ $v($ film $) / \mathrm{cm}^{-1} 1672,1620$ and 1600]; $\lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 219(\log \varepsilon$ 4.65), $248(\log \varepsilon 4.43)$ and $313(\log \varepsilon 3.84) ; \delta 1.46(6 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CMe}_{2}\right), 2.10\left(2 \mathrm{H}, \mathrm{t}, J 7,2-\mathrm{H}_{2}\right), 2.80\left(2 \mathrm{H}, \mathrm{t}, J 7,3-\mathrm{H}_{2}\right), 3.96(3 \mathrm{H}$, $\mathrm{s}, \mathrm{ArOMe}), 7.15(1 \mathrm{H}, \mathrm{d}, J 1,8-\mathrm{ArH}), 7.32(1 \mathrm{H}, \mathrm{dd}, J 8$ and 1 , 6-ArH), 7.56 ( $1 \mathrm{H}, \mathrm{d}, J 8,9-\mathrm{ArH}), 7.94(1 \mathrm{H}, \mathrm{d}, J 8,10-\mathrm{ArH})$ and $9.26(1 \mathrm{H} . \mathrm{d}, J 8,5-\mathrm{ArH})\left[\right.$ lit., ${ }^{16} \delta\left(\mathrm{CCl}_{4}\right) 1.4(\mathrm{~s}, 6 \mathrm{H}), 1.99(\mathrm{~m}$, $2 \mathrm{H}), 2.72(\mathrm{~m}, 2 \mathrm{H}), 3.83(\mathrm{~s}, 3 \mathrm{H}), 6.90-7.47(\mathrm{~m}, 3 \mathrm{H}), 7.73(\mathrm{~d}, 1 \mathrm{H}$, $J 8)$ and $9.10(\mathrm{~d} .1 \mathrm{H}, J 10)]$.

1,2,3,4-Tetrahydro-1,1-dimethylphenanthrene-4-one 6f.-The compound $4 f(300 \mathrm{mg}, 1.42 \mathrm{mmol})$ was converted in the same way as described for $\mathbf{6 a}$ into the title compound $\mathbf{6 f}(271 \mathrm{mg}, 85 \%$ ), b.p. ${ }^{150-155}{ }^{\circ} \mathrm{C}(0.1 \mathrm{mmHg})$ (Found: C, $85.75 ; \mathrm{H}, 7.35$. $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{O}$ requires $\mathrm{C}, 85.67 ; \mathrm{H}, 7.19 \%$ ); $v_{\max } / \mathrm{cm}^{-1} 1672,1627$ and $1600 ; \lambda_{\text {max }}(\mathrm{EtOH}) / \mathrm{nm} 219(\log \varepsilon 4.52), 248(\log \varepsilon 4.30)$ and $315(\log \varepsilon 3.86) ; \delta 1.49\left(6 \mathrm{H}, \mathrm{s}, \mathrm{CMe}_{2}\right), 2.14(2 \mathrm{H}, \mathrm{t}, J 7$, $\left.2-\mathrm{H}_{2}\right), 2.90\left(2 \mathrm{H}, \mathrm{t}, J 7,3-\mathrm{H}_{2}\right), 7.52-7.74(3 \mathrm{H}, \mathrm{m}, \mathrm{ArH}), 7.87(1 \mathrm{H}$, br d, 10-ArH), $8.06(1 \mathrm{H}, \mathrm{d}, J 8,9-\mathrm{ArH})$ and $9.33(1 \mathrm{H}, \mathrm{d}, J 8$, 5-ArH).

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